

JS 9/13/21

RJA 2/17/22

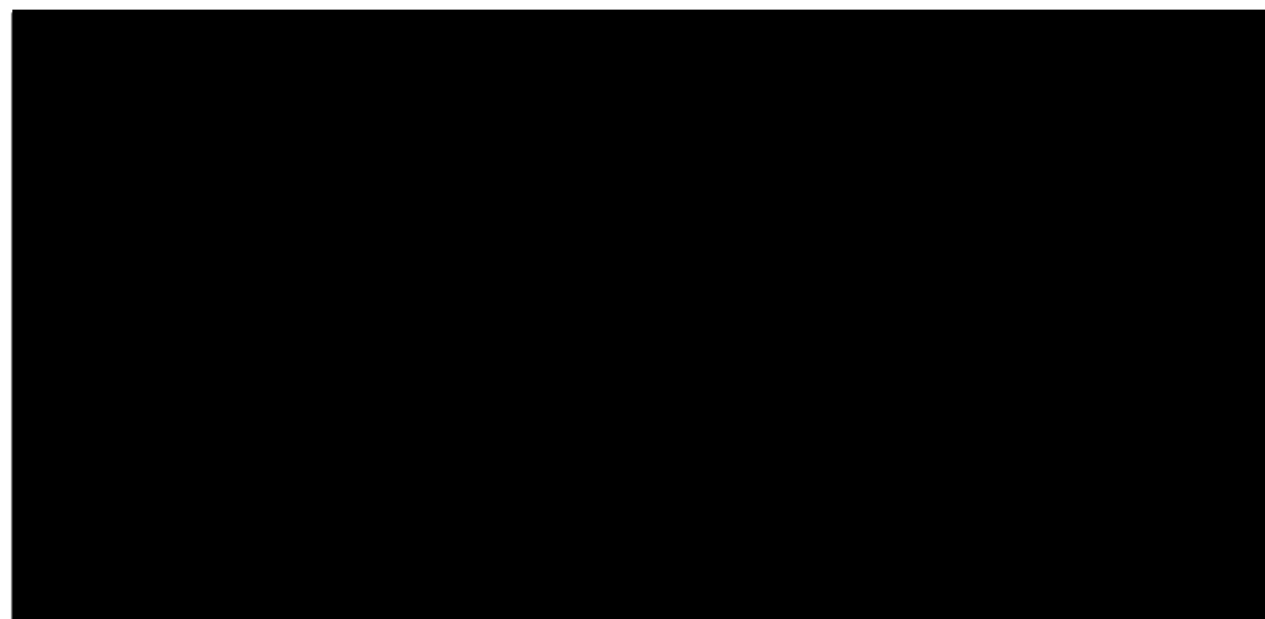
## Chemistry Report for Case # P-18-0284

### General

<b>Submitter:</b> [REDACTED]	
<b>Contact:</b> [REDACTED]	<b>Contact Telephone No.:</b> [REDACTED]
<b>TS No.:</b> UV47X9	
<b>Chemist:</b> Lin, D.	<b>Contractor Support:</b> Y
<b>PV Init (kg/yr):</b> [REDACTED]	<b>PV Max (kg/yr):</b> [REDACTED]
<b>Binding Option:</b> <input type="checkbox"/>	<b>Exposure-Based Review:</b> <input checked="" type="checkbox"/>
<b>Manufacture:</b> [REDACTED]	<b>Import:</b> <input type="checkbox"/>

**CAS Number:**None**Chemical Name** [REDACTED]  
[REDACTED]**Trade Name:** [REDACTED]**IES Order:** [REDACTED]**Generic Name:**Inorganic acid,  
reaction products with alkyl alcohol

### Chemical Structure



### Physical Chemical Properties

<b>Molecular Formula:</b> [REDACTED]	<b>Molecular Weight:</b> [REDACTED]
<b>% &lt; 500:</b>	<b>% &lt; 1000:</b>
<b>MP:</b>	<b>MP Estimate:</b>
<b>BP:</b>	<b>BP Pressure:</b>
<b>BP Estimate:</b> >500	
<b>VP (Torr):</b>	<b>VP Estimate (Torr):</b> <0.000001
<b>Water Solubility (g/L):</b>	<b>Water Soluble Estimate (g/L):</b> <0.000001/Reacts
<b>Log P:</b>	<b>Log P Estimate:</b> 22.41
<b>Physical State — Neat:</b> [REDACTED]	<b>Physical State — Manuf:</b> Solution: 90-93% PMN substance with [REDACTED]

<b>Physical State — Processing:</b> Solid blend: [REDACTED] % PMN substance
<b>Physical State — End Use:</b> Solid blend: [REDACTED] % PMN substance in plastic composite pellets

### Additional Chemical Info

The submitter provides the following composition: [REDACTED]

[REDACTED] The MF, MW and estimated values above are for the top triester.

Submitted Data: Light yellow [REDACTED]; WS < 10 g/L (MSDS); density = 0.97 g/cm<sup>3</sup>.

Estimated Data for top triester [EPI with MP =

20°C, MF = [REDACTED] MW = [REDACTED],

[REDACTED]  
BP = 694.57°C; VP = 1.26E-12 torr; WS = 3.30E-21 g/L; log P = 22.41.

Estimated Date for bottom diester [EPI with MP = 20°C, MF = [REDACTED]

[REDACTED], MW = [REDACTED], [REDACTED]: BP = 539.62°C; VP = 1.13E-11 torr; WS = 3.07E-13 g/L; log P = 14.77.

The PMN

is expected to hydrolyze with a half-life of days to give [REDACTED]

## Uses

**Consumer Use?** No

**Use:**

**Other Uses:**

Analogues

; Analogue [REDACTED]

## Reaction Description

## Pollution Prevention Analysis(P2 Analysis:)

None.

## Analogs

Analogues: [REDACTED]
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### Comments/Telephone Log

Artifact	Update/Upload Time
[REDACTED]	[REDACTED]